

# Highly Regio- and Diastereoselective Synthesis of CF<sub>3</sub>-Substituted Lactones *via* Photoredox-Catalyzed Carbolactonization of Alkenoic Acids

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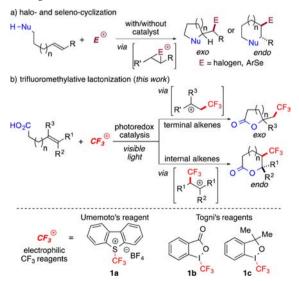
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Supporting Information

**ABSTRACT:** Trifluoromethylative lactonization of both terminal and internal alkenoic acids by photoredox catalysis has been developed. The use of a Ru photocatalyst and Umemoto's reagent as a CF<sub>3</sub> source is key in the present carbolactonization. This is the first example of a highly *endo-* and diastereoselective synthesis of CF<sub>3</sub>-substituted five-, six-, and seven-membered ring lactones from internal alkenoic acids.

Electrophile-involved cyclization of alkenes bearing a nucleophilic pendant is one of the most important reactions in organic chemistry and has been used for the synthesis of biologically relevant molecules. In particular, the reaction of internal alkenes is fascinating from the viewpoint of construction of two stereogenic centers via a single step. Strongly electrophilic reagents such as halogen and selenium reagents induce efficient halo- and seleno-cyclization of internal alkenes, and their exo/endo- and stereoselectivity is variable depending on the steric and electronic factors of both alkene and nucleophilic pendants as well as the catalyst used (Scheme 1a). In contrast, related efficient and selective carbocyclization is still under development; especially, carbolactonization of internal alkenoic acids is extremely limited. Thus, the

## Scheme 1. Electrophile-Involved Cyclization of Alkenes with a Nucleophilic Pendant



development of a novel strategy for carbolactonization is highly valuable for modern synthetic organic chemistry.

Since 2008, photoredox catalysis with well-defined ruthenium(II) polypyridine complexes (e.g., [Ru(bpy)<sub>3</sub>]<sup>2+</sup>) and the relevant cyclometalated iridium(III) derivatives has become a versatile synthetic tool because these compounds can readily catalyze single-electron transfer (SET) processes under visible light irradiation.<sup>4</sup> We are interested in photoredoxcatalyzed radical trifluoromethylation<sup>5,6</sup> using shelf-stable electrophilic trifluoromethylating reagents (+CF3) such as Umemoto's reagent 1a and Togni's reagents 1b and 1c<sup>7</sup> because the trifluoromethyl (CF<sub>3</sub>) group is a prevailing structural motif in many biologically active molecules as well as functional materials.<sup>8,9</sup> Despite significant progress in the synthesis of organofluorine compounds, 10 new methods for incorporation of a CF<sub>3</sub> group into diverse organic skeletons are very desirable. Recently, our group reported the highly regioselective trifluoromethylative difunctionalization of alke $nes^{3b,c,11}$  through a putative  $\beta$ -CF<sub>3</sub>-substituted carbocation intermediate generated by photoredox catalysis.<sup>6</sup> Based on these results, we envisaged photoredox-catalyzed trifluoromethylative lactonization of alkenoic acids with "+CF3" reagents (Scheme 1b).

Herein we disclose a novel protocol for regiocontrolled carbolactonization of both terminal and internal alkenoic acids by photoredox catalysis. It is remarkable that photoredox-catalyzed trifluoromethylative lactonization of internal alkenoic acids bearing a variety of functional groups proceeds with high *endo-* and diastereoselectivity under mild reaction conditions.

We initially tested the photocatalytic trifluoromethylative cyclization of 4-phenyl-4-pentenoic acid (2a) with Umemoto's reagent 1a using 5 mol % of  $[Ru(bpy)_3](PF_6)_2$  in  $CD_3CN$  under visible light irradiation (blue LEDs:  $\lambda_{max} = 425$  nm) for 2.5 h. As a result, trifluoromethyllactonization proceeded in an

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exo-selective manner to give the dihydrofuranone derivative with 2,2,2-trifluoroethyl substituent 3a in 86% NMR yield (Table 1, entry 1). The choice of <sup>+</sup>CF<sub>3</sub> reagents turned out to

Table 1. Trifluoromethylative Lactonization of 4-Phenyl-4pentenoic Acid  $2a^a$ 

entry	CF <sub>3</sub> source	base	solvent	% yield of $3a^b$
1	1a	none	CD <sub>3</sub> CN	86
2	1b	none	CD <sub>3</sub> CN	37
3	1c	none	CD <sub>3</sub> CN	69
4	1a	$K_2HPO_4$	$CD_3CN$	96
5	1a	$K_2CO_3$	$CD_3CN$	95
6	1a	$K_2HPO_4$	acetone- $d_6$	99
7	1a	$K_2HPO_4$	$dmso-d_6$	69
8 <sup>c</sup>	1a	$K_2HPO_4$	acetone- $d_6$	0
$9^d$	1a	$K_2HPO_4$	acetone- $d_6$	0
	_			

<sup>a</sup>Reaction conditions: A mixture of [Ru(bpy)<sub>3</sub>](PF<sub>6</sub>)<sub>2</sub> (1.25 μmol, 5 mol %), **1** (28 μmol, 1.1 equiv), and **2** (25 μmol, 1.0 equiv) in solvent (0.5 mL) was irradiated by 3 W blue LEDs ( $\lambda$  = 425 ± 15 nm) at room temperature. <sup>b</sup>Yields were determined by <sup>1</sup>H NMR spectroscopy. <sup>c</sup>In the dark. <sup>d</sup>No photocatalyst.

be crucial for the yield of 3a. Togni's reagents 1b and 1c gave 3a in low yields (entries 2 and 3). We found that the addition of a base such as  $K_2HPO_4$  and  $K_2CO_3$  resulted in a clean reaction, affording better yields of 96% and 95%, respectively (entries 4 and 5). Acetone and acetonitrile solvents are suitable for the present photocatalytic reaction (entries 6 and 7). Notably, product 3a was obtained neither in the dark nor in the absence of a photocatalyst (entries 8 and 9).

The results of *exo*-selective trifluoromethylative cyclization for various terminal alkenoic acids **2** are summarized in Table 2.

Table 2. Scope of the Present exo-Trifluoromethylative Lactonization of Terminal Alkenoic Acids $^{a,b}$ 

entry	n	R	% yield of product
1	1	Ph	3a: 90
2	1	$p ext{-MeOC}_6 ext{H}_4$	<b>3b</b> : 83
3	1	$p ext{-}FC_6H_4$	3c: 82
4	1	p-ClC <sub>6</sub> H <sub>4</sub>	3d: 79
5	1	p-BrC <sub>6</sub> H <sub>4</sub>	<b>3e</b> : 76
6	2	Ph	<b>3f</b> : 69 <sup>c</sup>
$7^d$	1	Н	3g: $60^{c}$

<sup>a</sup>Reaction conditions: see Supporting Information. <sup>b</sup>Isolated yields. <sup>c</sup>Yields were determined by <sup>1</sup>H NMR spectroscopy. <sup>d</sup>**2g/1a** = 1.5.

Reactions of 4-aryl-4-pentenoic acid derivatives with electron-donating (OMe: 2b), electron-withdrawing (F: 2c), and halogen (Cl and Br: 2d and 2e) groups on the benzene ring afforded the corresponding five-membered ring lactones in 83% (3b), 82% (3c), 79% (3d), and 76% (3e) yields, respectively. The present photocatalytic system can be extended to the construction of a six-membered ring lactone (3f) in good

yields. It should be noted that the present *exo*-selective carbolactonization is also applicable to unactivated 4-pentenoic acid (2g).

Next, we investigated photoredox-catalyzed trifluoromethylative cyclization of internal alkenoic acids 4 (Scheme 2).

Scheme 2. Scope of the Present *endo*-Trifluoromethylative Lactonization of Internal Alkenoic  $Acid^{a-c}$ 

<sup>a</sup>Reaction conditions: see Supporting Information. <sup>b</sup> Isolated yields. <sup>c</sup> The *trans/cis* ratios were determined by <sup>1</sup>H and <sup>19</sup>F NMR spectroscopies of crude reaction mixtures. <sup>d</sup> Acetone was used as solvent. <sup>e</sup> *trans/cis* = >20:1. <sup>f</sup> 2.0 equiv of 2,6-lutidine were used in place of  $K_2CO_3$ . <sup>g</sup> *trans/cis* = 8.9:1. <sup>h</sup> *trans/cis* = >49:1. <sup>i</sup> *trans/cis* (C3, C5) = 1:1. <sup>j</sup> *trans/cis* = >35:1.

Remarkably, the reaction of (E)-4-phenyl-3-butenoic acids (4a) proceeded with high *endo*- and diastereoselectivity to give  $(4R^*,5R^*)$ -4-trifluoromethyl-5-phenyldihydrofuran-2(3H)-one (5a) in 82% isolated yield as a single isomer. The reaction of (E)-4-aryl-3-butenoic acid derivatives with a methyl substituent, Me (4d), a methoxy group, MeO (4b), and a methylenedioxy substituent (4c) on the benzene ring produced diastereoselectively the corresponding 5-*endo* cyclized products 5b-d in high yields (72-79% yields). In addition, this reaction was amenable to internal alkenes bearing halogen atoms, F (4e), Cl (4f), Br (4g), and I (4h), and the  $CF_3$ -substituted fivemembered ring *endo*-lactones (5e-h) were obtained in high yields (72-83% yields) as a single diastereomer. A strongly

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electron-withdrawing group such as a trifluoromethyl group,  $CF_3$  (4i), did not hinder the reaction. Moreover, alkenoic acid with benzothiophene (4j), a heteroarene, was also tolerated and the 5-endo cyclization proceeded with exclusively transselectivity to give the  $CF_3$ -substituted lactone (5j) in 41% yield. The structure of 5j was unequivocally confirmed by single-crystal X-ray analysis (Figure 1).<sup>12</sup>

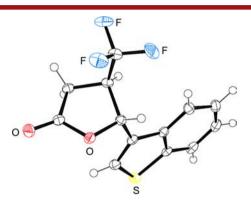


Figure 1. Single-crystal X-ray analysis of 5j. The thermal ellipsoids are set at a 50% probability level.

The present 5-endo cyclization was extended to the reaction of 1,1,2-trisubstituted internal alkenoic acids. The trifluoromethyllactonization of 4,4-diphenyl-3-butenoic acid (4k) gave the endo-lactone (5k) with a quaternary carbon center in 81% yield. In the reaction of (E)-4-phenyl-3-pentenoic acid (4l), only the trans-isomer with the diastereoselective formation of a quaternary carbon center (5l) was obtained in 85% yield. It is noteworthy that the reaction of  $\beta$ -cyclohexylidenepropionic acid (4m) afforded the spirocyclic endo-lactone product (5m) in 53% yield, indicating that this trifluoromethylative lactonization can be applied to an alkyl-substituted alkenoic acid as well as an aryl-substituted one.

Furthermore, the present protocol enables endo- and diastereoselective synthesis of CF3-substituted six- and sevenmembered ring lactones. The reaction of (E)-5-phenyl-4pentenoic acid (4n) proceeded with 6-endo cyclization to produce the CF<sub>3</sub>-substituted six-membered ring lactone (5n) in 52% yield with a high level of diastereocontrol (8.9:1 dr). Notable improvements in yield and diastereoselectivity were observed when the alkenoic acids bearing dialkyl substituents at the  $\alpha$ -position of the carboxyl group (40-q) were used. Indeed, the 6-endo cyclized products (5o-q) were obtained in good yields (67-78%) with excellent diastereoselectivities (trans/cis (C5, C6) = >49:1 dr). The structure of **50** was determined by X-ray analysis (see Supporting Information).<sup>12</sup> In addition, the present method was also applicable to the synthesis of the CF3-substituted seven-membered ring endolactone (5r). These results show that photoredox-catalyzed trifluoromethyllactonization of both terminal and internal alkenoic acids allows us to easily access a variety of lactones containing a CF<sub>3</sub> group in a regioselective manner. In particular, it is worth noting that highly endo- and diastereoselective carbolactonization of internal alkenoic acids is an unprecedented transformation to the best of our knowledge.

Control of stereogenic centers is the next stage for this work. We tested the diastereocontrolled reaction utilizing the adjacent substituent. The reaction of (S)-4-phenyl-2-methyl-3-butenoic acid (4s) afforded the  $CF_3$ -substituted *endo*-lactones (5s) in 60% yield as a mixture of only two of four possible

diastereomers (3.3:1 dr) with a moderate level of stereocontrol (eq 1). Lowering the reaction temperature to -78 °C improved

the diastereoselectivity (cf. 2.4:1 at rt). This result implies that the facial-selective introduction of the CF<sub>3</sub> group is key for the stereoselective reaction.

To gain insight into the mechanism, we conducted the photocatalytic cyclization of (Z)-alkenoic acid ((Z)-4a) (eq 2).

The reaction provided a product identical to the one obtained from the (E)-isomer 4a. This *trans*-selective formation strongly supports the two isomeric substrates producing the same intermediate.

The reaction appears to proceed through a mechanism similar to that of the previously reported oxy- and aminotrifluoromethylation of alkenes by photoredox catalysis. 6a,c Sequential SET photoredox processes in the course of the reaction of <sup>+</sup>CF<sub>3</sub> with alkenoic acids 2 and 4 by [Ru(bpy)<sub>3</sub>]<sup>2+</sup> under visible light irradiation produces a  $\beta$ -CF<sub>3</sub>-substituted carbocation intermediate in a highly regioselective manner through radical transformation (for the fully proposed mechanism, see Supporting Information). Nucleophilic attack of the dangling carboxylic acid affords the CF3-substituted lactone product. To elucidate high diastereoselectivity, we performed DFT calculations for possible conformations of a generable  $\beta$ -CF<sub>3</sub> carbocation from 4a. It was revealed that, in the most favorable conformation, the CF<sub>3</sub> group and the phenyl group are arranged in an anti-fashion presumably because of steric repulsion between the CF<sub>3</sub> and the Ph groups (see Supporting Information). Subsequent intramolecular nucleophilic addition of the COOH group to the carbocationic center gives the trans-product, which is actually obtained by the experiments (Scheme 3).

In conclusion, we have developed trifluoromethylative lactonization of both terminal and internal alkenoic acids by photoredox catalysis under mild conditions. Regioselective trifluoromethylation of C=C bonds and intramolecular

Scheme 3. A Plausible Reaction Mechanism for High Diastereoselectivity Observed for 5a

HO<sub>2</sub>C

(E)-4a

Ph

photoredox

catalysis

$$CF_3$$
 $F_3$ 
 $F_4$ 
 $F_5$ 
 $F_7$ 
 $F_7$ 

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nucleophilic attack of a pendant carboxylate to the  $\beta$ -CF<sub>3</sub> carbocation intermediate achieves highly regiocontrolled carbolactonizaton of alkenoic acids. In particular, this operationally easy trifluoromethylative cyclization protocol for internal alkenoic acids enables us to access a variety of CF<sub>3</sub>-substituted five- and six-membered ring *endo*-lactones bearing many functional groups in a diastereoselective manner. Furthermore, this photocatalytic cyclization can be applied successfully to the synthesis of a CF<sub>3</sub>-substituted seven-membered ring *endo*-lactone. This is the first example of *endo*-carbolactonization of internal alkenoic acids. Further studies on stereoselective trifluoromethylation are underway in our laboratory.

#### ASSOCIATED CONTENT

### Supporting Information

Details of experimental procedures and full spectroscopic data for all new compounds, crystallographic data of 5j (CCDC 962013) and 5o (CCDC 962014), and results of computational analysis. This material is available free of charge via the Internet at http://pubs.acs.org.

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#### Notes

The authors declare no competing financial interest.

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